

# Prospects in sustainable lead pollution management: A microbiological perspective.

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## Abstract

Lead pollution is a major biotic and abiotic stress with significant impact on environmental, public health, agriculture and a consequent threat to food security. Hence, the need for effective and sustainable remediation strategy is strongly required to remove lead pollutant and restore environmental integrity. In this study, four lead tolerant bacteria *Bacillus infantis* strain K66, *Halopseudomonas Xiamenensis* strain B13, *Lysinibacillus fusiformis* strain KAF67 and *Pseudomonas* spp. strain A27 were used for lead bioremediation. The isolates were screened for the presence of lead tolerance gene clusters PbrA,B,C and T encoding P-type Pb(II) efflux ATPase; predicted integral membrane protein; predicted prolipoprotein signal peptidase and Pb(II) uptake protein respectively. All isolates harbored three genes while KAF67 harbored all. The isolates were utilized for lead treatability study under greenhouse conditions using a block randomized design system for 56 days. Lead bioremediation was monitored biweekly using bacteria counts and lead concentration as monitoring indices. Results revealed a decrease in lead concentration across all pots however, there was a significant difference ( $p \leq 0.05$ ) between the lead bacteria treated pot and control. The rate of removal of lead was highest in pot amended with K66 (84.64%) and lowest in control pot (40.91%). The combined bacterial strains were effective in the remediation of the lead stressed soil.

**Keywords:** Heavy metal, Bioremediation, Lead, Bacteria, Molecular mechanism, Lead toxicity, PGPR.

## Introduction

Various anthropogenic activities have caused environmental contamination to reach its apex. Quarrying, as every mining operation, is a destructive development activity whose socio-economic benefits may be unable to compensate for the overall detrimental effects on natural ecosystems. Blasting and crushing of rocks and use of explosives and heat releases particulate matter and dust of different metallic constituents from the machineries and blasting processes. Through this process, a large quantity of heavy metals is released into the environment as dust particles, especially into surrounding soils [1]. The most prevalent environmental pollutant and very harmful substance is heavy metal [2]. A few heavy metals occur naturally in soil but can however cause major harm when they are in higher concentrations. Pollutions caused as a result of increased concentrations of heavy metals have become a concern and threat to the environment and living organisms [3]. There is an urgent need for a remediation solution for heavy metal contamination, which is a significant environmental issue that affects the entire planet [4].

Lead (Pb) pollution has gotten the most attention of all the heavy metals due to its widespread industrial use and exceedingly hazardous properties. Lead is very valuable and easily available, which causes a lot of lead-containing waste

and puts all living creatures in danger. Lead was put on the environmental potential agency agenda and 20-26 October was designated as National Lead Poisoning Prevention Week (NLPPW). The National Lead Poisoning Prevention Week (NLPPW) seeks to unite people, groups, businesses, states and local governments in order to raise awareness of the health risks posed by lead pollution and to take action to reduce lead exposure to humans and other living things.

Common physical and chemical techniques have failed to eliminate lead pollution in a way that is long-lasting. As a result, attention has been drawn to bioremediation methods that provide long-term remedies for lead pollution. Utilizing biological processes, bioremediation cleans up pollution in an efficient manner without adding more pollution. Bioremediation can be defined as a process of employing any biological entity to remediate pollutants. It can be achieved with the help of bacteria, fungus, algae, plants or any part of a living being like enzyme, or exopolysaccharide. Among all living organisms, bacteria are widely recognized for their bioremediation efficacy due to versatility and adaptability. Bacteria are capable of growing in adverse conditions by developing adaptable mechanisms. Several bacterial species such as *Acinetobacter*, *Pseudomonas*, *Bacillus*, *Gluconacetobacter*, and *Serratia* have been reported to solubilize and/or tolerate lead. However, reports

on the ability of *Bacillus infantis* strain K66, *Halopseudomonas xiamenensis* strain B13, *Lysinibacillus fusiformis* strain KAF67 and *Pseudomonas* spp. strain A27 to bioremediate lead polluted soils are scarce.

In this study, *Bacillus infantis* strain K66, *Halopseudomonas xiamenensis* strain B13, *Lysinibacillus fusiformis* strain KAF67 and *Pseudomonas* spp. strain A27 have shown to be lead resistant bacteria. Therefore, the present work was designed to study the ability of lead-resistant bacteria in alleviating lead pollution in soil.

## Materials and Methods

### ***Bacteria collection and culture condition***

The *Bacillus infantis* strain K66, *Halopseudomonas xiamenensis* strain B13, *Lysinibacillus fusiformis* strain KAF67, and *Pseudomonas* spp. strain A27 were previously isolated from the contaminated soil of Okpella mining site, screened for lead resistant potential, plant growth promoting traits and identified based on 16S rRNA sequencing and were preserved in 20% glycerol stock at 4°C and were refreshed in nutrient agar for 24 hours at 37°C before use in this study.

### ***Functional gene analysis of lead resistant organisms***

DNA extraction and high-throughput sequencing Bacterial community succession in soil samples during remediation were determined. Briefly, the total DNA in samples collected on days 1, 7, and 35 were extracted with the FastDNA<sup>®</sup> SPIN Kit for soils (Mpbio, USA) according to manufacturer's protocols. The V3-V4 hypervariable regions of the 16S rRNA gene were amplified with the primers PbrA, PbrB, PbrC, and PbrF. After purification, quantification, and pooling, a clone library was constructed using the TruSeq<sup>®</sup> DNA PCR-Free sample preparation kit (Illumina, USA) and the amplified DNA samples were sequenced by Illumina NovaSeq 6000 (Novogene, Beijing). All the raw sequence data was filtered using Qiime quality filters (Version, 1.9.1) to remove the low-quality sequence reads. After that, the remaining sequences were clustered by using Uparse software (Uparse v. 7.0.1001) as assigned to Operational Taxonomic Units (OTUs) at similarities of 97%. Alpha diversity was applied in analyzing the complexity of species diversity through Chao1, Shannon, and Simpson indices, and beta diversity analysis was conducted to evaluate the difference of samples in species complexity using Qiime software. Additionally, Un-weighted Pair-Group Method with Arithmetic means (UPGMA) Clustering was performed as a type of hierarchical clustering method to interpret the distance matrix using average linkage and was conducted by QIIME software (Version, 1.9.1).

### ***Collection of soil sample***

For the purpose of collecting soil samples, Neboh et al. methodology was used. The samples were taken early in the day, between 9:00 and 11:00 am, when activity was at its height. To allow for the best bacterial activity, the samples of soil were dug out at a depth of 0 to 30 cm and collected using

aluminum foils for both an organic chemical analysis and a microbiological examination. In ice jackets, the soil samples were delivered to the lab at Edo State University while the maize seedling was collected from ministry of agriculture.

### ***Soil physiochemical analysis***

Soil pH value and Electric Conductivity (EC) was measured at a soil/water mass ratio of 1:2.5 by using a pH meter and conductance meter, respectively. Soil Organic Matter (SOM) content was determined via K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> oxidation and FeSO<sub>4</sub> titration. The water extractable organic carbon in soils was extracted with deionized water at a 1:20 (w/v) and measured with a total carbon analyzer (TOC-VCSH, Shimadzu, USA). Additionally, the available P, available K, and inorganic N (NO<sub>3</sub> and NH<sub>4</sub>) in soils were analyzed using UV Spectrophotometer (UVmini-1240, Shimadzu, Japan), flam spectrophotometer (6400A, Shjingmi Inc. China) and flow injector auto-analyzer (AA3. SEAL Analytical Inc. USA), respectively, after extraction. In terms of metals, the total Cd content was acid digested with a mixture of HCl-HNO<sub>3</sub>-HClO<sub>4</sub>, according to NY/T1613-2008 (MOA, 2008), and the available lead were extracted with Diethylenetriamine Pentaacetic Acid (DTPA) solution (1:10, w/v; pH ¼ 7.3). The lead contents in digestant and leaching solutions were measured using a flame atomic absorption spectrometry (FAAS, PerkinElmer, Analyst 700, USA) [5].

### ***Total culturable heterotrophic species count***

The soil sample was weighed at 1 g into a beaker and mixed with 9 ml of distilled water homogenously and test tubes were arranged in test tube rack, 9 ml of pure water was measured into the test tubes and 1 ml of mixed sample was measured from the beaker into the test tubes one with the aid of syringe and from test tube one, 1 ml was measured into test tube two and from two to three until the last fold [6]. The soil samples after serial dilution were suspended in nutrient agar. Plate was hatched at 37°C for 24 hours and colonies with morphological characteristic were counted [6].

### ***Experimental design for bioremediation of lead contaminated soil***

Bioremediation study was carried out in a greenhouse condition using block randomized design for the experiment. Contaminated (spiked soil) and uncontaminated (control) soil samples were used in the bioremediation investigation. The experiment consists of five treatments; pot 1 treated with *Bacillus infantis* strain K66, pot 2 treated with *Halopseudomonas xiamenensis* strain B13, pot 3 treated with *Lysinibacillus fusiformis* strain KAF 67, pot 4 treated with *Pseudomonas* spp., pot 5 without bacteria (control experiment). 10% (v/v) of microbial inoculum was added to 5 g of contaminated soil. Inoculums for each treatment contained 3 × 10<sup>9</sup> CFU/mL. The experimental set up lasted for 56 days and lead bioremediation was monitored biweekly using bacteria count and lead concentration reduction respectively. The soil

was irrigated with 250 ml of Water to keep the moisture level at 60-65% [7].

### **Spiking of soil with lead solution**

In a laboratory, a solution of binary metal salt (PbCl) was used to intentionally contaminate the soil. Zero point five grams (0.5 g) of PbCl salt was incorporated into 250 ml of distilled water to achieve a concentration of 500 mgL<sup>-1</sup> of lead and was added to the soil sample. The level of lead in the spiked soils was based on the “Soil environmental quality-risk control standard for soil contamination of agricultural land (GB15618-2018)” (MEEPRC, 2018), where risk management value is 0.5 mgL<sup>-1</sup>.

### **Establishing a microbial consortia for lead bioremediation in polluted soil**

Prior to being inoculated into nutrient broth and cultured for 24 hours, individual strains were initially cultivated in nutrient agar for 24 h at 37°C in automatic orbital shaker fixed phase at 150 rpm. The various strains were pooled out in equal proportions at a wavelength of 600 nm after reaching a growth of 1.3 ABS [8-10].

**Table 1.** Functional genes clusters present in lead resistant bacteria.

Organisms	Primers			
	PbrA	PbrB	PbrC	PbrT
<i>Halopseudomonas xiamenensis</i>	+	+	+	-
<i>Lysinibacillus fusiformis</i>	+	+	+	+
<i>Pseudomonas</i> spp.	+	+	+	-
<i>Bacillus infantis</i>	+	+	+	-

### **Base line physiochemical parameters of the sample soil**

The base line result of the soil physiochemical properties revealed that the lead concentration in the soil was above permissible level by WHO and also other parameters were at

**Table 2.** Base line physicochemical data of lead polluted soil sample.

Parameters	Soil sample	Normal soil
Moisture content (%)	1.23	10-14
pH	8.44	5.8-6.2
Electrical conductance (mS/cm)	47	0.8-2.4
Organic carbon (%)	0.58	02-10
Organic matter (%)	0.96	03-6
Phosphate (mg/kg)	75.16	25-50
Nitrate (mg/kg)	4.06	10-50
Pb (mg/kg)	40.86	0-17

### **Base line of total culturable heterotrophic and lead resistant bacteria count**

The result revealed that the lead resistant bacteria population was higher than the total culturable heterotrophic bacteria

### **Statistical analysis**

Results from this study are shown as means with Standard Deviations (SD). The difference between many therapies was done utilizing one-way Analysis of Variance to assess (ANOVA), and Duncan's multiple comparisons were performed to ascertain the variation's relevance between various treatments at p<0.05 [11]. Utilizing origin 2020, the data were plotted. The Bioaccumulation Factors (BCF) that come after the following equations were used to quantify the transport of lead from soil to maize plants:

$$BCF^{1/4} M_1=M_2$$

Where M<sub>1</sub> represents the lead content in maize plants (mg kg<sup>-1</sup> dry weight) and M<sub>2</sub> represents the metal occurrence in soil (mg kg<sup>-1</sup> dry weight).

### **Results**

The results show in Table 1.

variance which is an indication of the consequence of lead pollution (Table 2).

which underscore a long term input of lead in the soil (Table 3).

**Table 3.** Base line count of heterotrophic bacteria and lead-resistant bacteria.

Total heterotrophic bacteria	Lead resistant bacteria
$6.5 \times 10^5$	$2.5 \times 10^7$

**Monitoring during bioremediation using microbiology indices**

Monitoring during the bioremediation across different days revealed that there was a decrease in the bacteria population both the total heterotrophic bacteria and lead resistant bacteria

as the time progresses. This could be as a result of environmental conditions (Table 4).

**Table 4.** Mean values of bacterial count across different monitoring days for THB and LRB.

Isolates	Day 0		Day 14		Day 28		Day 42		Day 56	
	THB	LRB	THB	LRB	THB	LRB	THB	LRB	THB	LRB
Control	$2.6 \times 10^7$	$3.3 \times 10^7$	$2.2 \times 10^7$	$3.7 \times 10^7$	$1.1 \times 10^7$	$2.0 \times 10^7$	$1.3 \times 10^7$	$2.1 \times 10^7$	$3.7 \times 10^6$	$1.3 \times 10^7$
Pbr1	$1.0 \times 10^8$	$3.0 \times 10^8$	$9.7 \times 10^7$	$2.4 \times 10^8$	$9.0 \times 10^7$	$1.6 \times 10^8$	$8.3 \times 10^7$	$1.8 \times 10^8$	$7.1 \times 10^7$	$1.5 \times 10^8$
Pbr2	$1.5 \times 10^8$	$3.2 \times 10^8$	$1.0 \times 10^8$	$2.6 \times 10^8$	$8.4 \times 10^7$	$2.0 \times 10^8$	$7.8 \times 10^7$	$2.9 \times 10^8$	$6.5 \times 10^7$	$1.6 \times 10^8$
Pbr3	$9.6 \times 10^7$	$1.9 \times 10^8$	$9.0 \times 10^7$	$1.8 \times 10^8$	$7.7 \times 10^7$	$1.5 \times 10^8$	$7.6 \times 10^7$	$1.5 \times 10^8$	$6.7 \times 10^7$	$1.4 \times 10^8$
Pbr4	$1.4 \times 10^8$	$2.1 \times 10^8$	$9.6 \times 10^7$	$1.4 \times 10^8$	$9.7 \times 10^7$	$1.7 \times 10^8$	$9.0 \times 10^7$	$1.9 \times 10^8$	$7.8 \times 10^7$	$1.5 \times 10^8$

**Note:** THB: Total Heterotrophic Bacteria; LRB: Lead Resistant Bacteria

**Spiking of soil with lead solution**

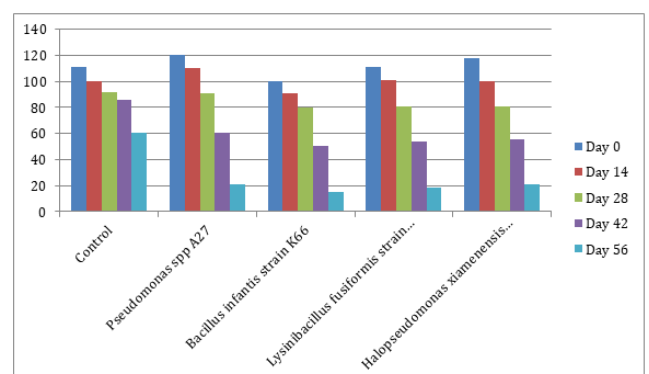
The soil sample was spiked using lead solution to increase the lead concentration and also for an effective monitoring (Table 5)

**Table 5.** Lead concentration in the soil before and after spiking (mg/kg).

Treatments	Before	After
	(mg/kg)	(mg/kg)
Control	40.803	110.86
Pbr1	50.531	120.58
Pbr2	30.476	100.33
Pbr3	40.825	110.91
Pbr4	40.678	117.74

**Monitoring using lead concentration degradation across different days**

Monitoring of lead concentration degradation revealed that *Bacillus infantis* strain K66 had the highest percentage degradation which was closely followed by *Lysinibacillus fusiformis* strain KAF67, *Pseudomonas* spp. strain A27, and then *Halopseudomonas xiamenensis* strain B13 while control pot had the lowest lead percentage degradation (Figure 1).



**Figure 1.** Lead reduction profile across all pots from day 0-56.

**Lead percentage degradation during bioremediation**

The percentage of lead degradation revealed that *Bacillus infantis* strain K66 had the highest percentage degradation which was closely followed by *Lysinibacillus fusiformis* strain KAF67, *Pseudomonas* spp. strain A27, and then *Halopseudomonas xiamenensis* strain B13 while control pot had the lowest lead percentage degradation (Figure 2).

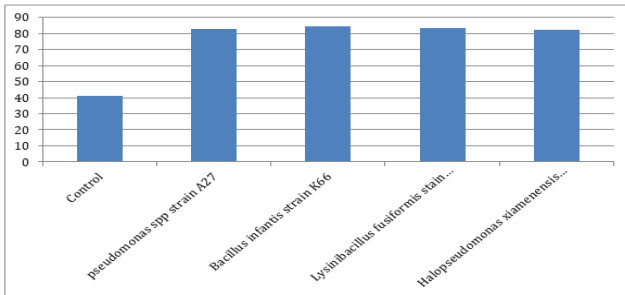


Figure 2. Percentage degradation of lead content.

Table 6. Shoot length of maize plant at day 7, 14, 21 and 28 (cm).

Days	Control pot (cm)	Pot1 (cm)	Pot2 (cm)	Pot3 (cm)	Pot4 (cm)
7	4	7	9	7	8
14	18	27	31	30	29
21	24	49	56	53	48
28	47	84	92	86	81
Standard deviation	17.91	32.99	35.62	33.71	30.99

**Root length of maize plant**

The root length of maize plant revealed that maize plant in pot 2 had the highest shoot length (20.09 cm), followed by pot 3

(18.71 cm), pot 1 (17.78 cm) and then pot 4 (16.82 cm) while the control pot had the shortest root length (8.74 cm) (Table 7).

Table 7. Root length of maize plant at day 7, 14, 21 and 28 (cm).

Days	Control pot (cm)	Pot1 (cm)	Pot2 (cm)	Pot3 (cm)	Pot4 (cm)
7	1.4	4	4	5	5
14	4	7	7	8	8
21	7	15	18	17	16
28	21	41	48	46	44
Standard deviation	8.74	16.82	20.09	18.71	17.78

**Fresh and dry root weight of maize plant**

It was observed that pot 2 had the highest fresh root weight which was (3.18 cm) and dry root weight (1.03 cm) of maize plant followed by pot 3 (3.07, 1.01 cm), pot 1 (2.84, 0.96 cm) and then pot 4 (2.76, 0.89 cm) while the control pot had the

lowest fresh and dry root weight (0.82, 0.25 cm) of the maize plant (Table 8).

Table 8. Fresh and dry root weight of the maize plant after cultivation (cm).

Treatment	Fresh root weight	Dry root weight
Control pot	0.82	0.25

Pot1	2.84	0.96
Pot2	3.18	1.03
Pot3	3.07	1.01
Pot4	2.76	0.89

### ***Chlorophyll content of maize plant***

It was observed that pot 2 had the highest chlorophyll content (63%) in the maize plant followed by pot 3 (60%), pot 1(59%)

and then pot 4 (58%) while the control pot had the lowest chlorophyll content (24%) (Table 9).

**Table 9.** Chlorophyll content of the maize plant (%).

Treatment	Percentage (%)
Control	24
Pot1	59
Pot2	63
Pot3	60
Pot4	58

### ***Lead uptake by maize plant***

It was observed that only the control pot had lead uptake of (48%) in the maize plant while lead uptake was not detected in

other pots after 21 days of cultivation (Table 10).

**Table 10.** Lead uptake in maize plant (%).

Treatment	Percentage (%)
Control pot	48
Pot1	not discovered
Pot2	not discovered
Pot3	not discovered
Pot4	not discovered

### ***Residual lead in soil after maize cultivation***

It was observed that only control pot had residual lead content

of (53%) present while other pots had no residual lead content at the end of the cultivation period (Table 11).

**Table 11.** Residual lead content present in soil after cultivation of maize (%).

Treatment	Percentage (%)
Control	53
Pbr1	not discovered
Pbr2	not discovered
Pbr3	not discovered
Pbr4	not discovered

## **Discussion**

This study involves the use of different strains of lead resistant bacteria combined with plant growth promoting bacteria; *Lysinibacillus fusiformis* strain BT3 to achieve the bioremediation and restoration of lead polluted soil. The lead

resistant bacteria were obtained from previous studies after screening using different parameters to assess their tolerance potential to lead and identified with the 16S rRNA typing method. In this study, the genetic potential of these organisms were further assessed to confirm their resistance capability using *PbrABC* gene clusters responsible for lead

bioprecipitation from the environment [12]. The lead polluted soil sample prior to bioremediation was analysed for physicochemical properties and subsequently treated with the different lead resistance bacteria for a period of 56 days with periodic monitoring of lead concentration and bacterial count. At the end of the treatability study, strain BT3 was introduced to the soil as biofertilizer for the cultivation of maize to support the restoration of the lead treated soil [13].

At the end of the functional gene analysis *Pseudomonas* spp. strain A27, *Bacillus infantis* strain K66, and *Halopseudomonas xiamenensis* B13 harbored three of the gene cluster *PbrA*, *PbrB* and *PbrC* while *Lysinibacillus fusiformis* strain KAF67 harbored all of the genes in the cluster: *PbrA*, *PbrB*, *PbrC* and *PbrT* ascertaining their potential to tolerate lead using the bioprecipitation mechanisms which has been established according to Utami et al. [14]. The lead resistant gene cluster *pbrTRABCD* in *Cupriavidus metallidurans* CH<sub>34</sub> has been implicated in the molecular mechanism of lead resistance in previous studies [15]. According to Sevvat et al., where *PbrT* encode a putative lead uptake which help to reduce the lead concentration in the soil, *PbrB/PbrC* creates integral membrane proteins having characteristics like signal peptidase responsible for phosphatase and *PbrA* produces an Adenosine Triphosphate (ATPase) transporter that aids in the cytoplasmic export of lead. As a result of the inorganic phosphate reaction produced by undecaprenyl pyrophosphate phosphatase, which is encoded by *PbrB*, the lead is further sequestered in precipitated form while to stop lead from being reintroduced into the microbes, *PbrA* transfers lead out of the cell, where it is precipitated by the inorganic phosphate that *PbrB* releases. *PbrC* creates essential membrane proteins which perform the role of signal peptidase responsible for phosphatase while *PbrT* encodes the potential lead uptake and lead-binding proteins. *PbrABC* also referred to as phosphate ATP-Binding Cassette (ABC) transporters have been established to be expressed by bacteria which aids in the extracellular production of phosphate needed to precipitate lead and these set of genes were present in strains A27, K66 and B13, respectively [16]. All of these genes in the cluster are connected and work together inside the bacterial cell to promote lead removal through the bioprecipitation mechanism. However, other kinds of molecular defenses are available to bacteria to deal with lead toxicity such as biosorption, efflux mechanism, siderophore production and others [17].

The physicochemical characteristics of the soil sample were at variance when compared with pristine soil. This served as baseline result to ascertain the prevailing inherent soil properties prior to bioremediation. In the soil sample, there was a greater level of lead compared with pristine soil which shows the effect of the mining activities in the area. According to Liu et al., crushed rock mining activities generates considerable amount of dust and wastes, which significantly increases the concentration of lead in the soil. These lead are mobilized or dissolved into the soil which tends to increase the concentration of the natural deposits and alter other physicochemical properties in the soil [18]. Different authors have pointed out the contribution of mining activities to the soil such as

increase in lead concentration, reduction of soil fertility and biodiversity, decreased output and plant growth and increase in soil alkalinity. The higher concentrations of lead recorded in the soil sample within the mining site also confirms the relationship between lead concentration and mining activities and the results generated from this study are in line with the work of Nwovu et al. Other parameters analyzed such as nitrate and phosphate showed alterations in the physicochemical structure when compared to pristine soil, also an evidence of the effect of long term lead pollution. The phosphate content was higher in the contaminated soil than the pristine soil while nitrate was lower in the polluted soil sample (Table 4). The result may look incorrect or inaccurate but as earlier stated microorganisms responsible for bioprecipitation as lead toxicity reduction mechanism make use of Phosphate-Solubilizing Bacteria (PSB) which play a crucial part in soil lead bioremediation by releasing the phosphate needed for the lead bioprecipitation reaction from insoluble phosphate molecules like Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> [19]. Now that phosphate is bioavailable, it can interact with lead to create an insoluble lead phosphate, which will make lead less mobile. Also, the THB population in the mining site is lower than LRB which infers a possible reduction in the nitrogen fixing bacteria and nitrifying bacteria population responsible for producing nitrates and other nitrogen-based derivatives as a result of the toxic effect of lead contamination, this could be the reason for the low nitrate content in the soil sample. According to Raghad et al., heterotrophic microbes found in contaminated soil were observed to be very low as this is due to the sensitivity of nitrifying bacteria and other non-lead resistant bacteria to high lead concentrations.

Since lead pollution is known to affect the biotic and abiotic components of the environment, bioremediation has been established to be the most effective ecofriendly and cost effective approach. In this study, the bioremediation of lead was achieved employing the lead tolerant bacteria as remediation agents in order to remove lead from the sample soil. At the end of the 56 days period, strain K66 caused the highest percentage of lead reduction which was closely followed by strain KAF67, A27 and B13, all samples treated under same experimental conditions. These organisms individually caused >80% lead reduction. According to Kuddus et al., bioremediation of heavy metals can be scored successful when ≥ 65% or more of the metals are removed from a polluted environment and this indicates that the lead resistant bacteria displayed highly proficient capacity to precipitate and remove lead from the sample. This was achieved based on the functional gene analysis which showed the bacteria harbored gene clusters responsible for bioprecipitation of lead from soil. As reported by researchers, some bacteria have a stronger affinity and sensitivity to lead than others, which may explain the variance in the amount of lead removed by each treatment. When compared to a similar study by Fauziah et al., which investigated remediation of lead-contaminated soil using microbe isolated from a closed dump site, the removal activities of lead in the soil samples were very high. They found that adding microbes, specifically, introducing proteobacteria to leachate-contaminated soil may significantly

reduce the heavy metal concentration, and adding bacterial groups to contaminated soil can remove metals from the environment more effectively. Although strain K66 harbored three of the gene clusters, it scored the highest remediation percentage which supports the fact bacteria responds to lead contamination differently, not just based on their genetic resistance signatures but also natural affinity and sensitivity. However, some *Bacillus* strains are known to be lead resistant with a high potential to resist a wide range of lead derivatives/lead-based compounds. According to Qiao et al., *Bacillus subtilis* X3 bioprecipitates lead in a variety of forms, including  $Pb_5(PO_4)_3OH$ ,  $Pb_{10}(PO_4)_6(OH)_2$ , and  $Pb_5(PO_4)_3Cl$  which supports the above statement.

No Significant Difference (SD) was recorded between the treatments however there was SD between treatment and control at  $p < 0.05$ . The control had the least recorded lead removal and this could be attributed to the absence of any treatment or inoculant to enhance and sustain the remediation process which could be due to the presence of indigenous LRBs.

The increased level of lead in the environment has significantly affected the overall microbial activity and community as well as population size. Several research projects, based on the isolation-based protocol applied, have demonstrated that lead contamination gave rise to shifts in microbial population which is consistent with the findings from this study [20]. As seen in Table 4, after introducing the bacteria into the soil during bioremediation, the results revealed that higher populations of culturable LRB were obtained compared to THB pre and during bioremediation across all treatments and sampling days which shows a modification or shift in the bacterial population as a result of lead contamination. Also, the control pot which was un-inoculated showed a higher count of LRB consistently throughout the treatability study although both populations decreased across different days of investigation. From this study, it implies that the bioremediation of lead in soil samples were very successful.

## Conclusion

Lead is a priority contaminant and highly hazardous. There have been numerous reports of lead pollution around the world, demonstrating the urgent need for environmentally friendly lead cleanup technologies. Lead resistance has been documented in a wide range of microorganisms. Different defense mechanisms have been established by lead-resistant bacteria to combat lead poisoning. Furthermore, based on several bacterial bioremediation methods, the potential applicability of lead-resistant bacteria was deduced. Combined application of lead-reducing bacteria holds great promise in gaining environmental sustainability.

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